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Chapter 16 Thermal Energy and Heat

Section 16.2 Heat and Thermodynamics

(pages 479-483)

This section discusses three kinds of thermal energy transfer and introduces the first, second, and third laws of thermodynamics.

Reading Strategy (page 479)

Build Vocabulary As you read this section, add definitions and examples to complete the table. For more information on this Reading Strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.

Transfer of Thermal Energy				
Definitions	Examples			
Conduction: transfer of thermal energy with no net transfer of matter	Frying pan handle heats up			
Convection:				
Radiation:				

Conduction (pages 479-480)

- **1.** The transfer of thermal energy with no overall transfer of matter is called ______.
- 2. Why is conduction slower in gases than in liquids and solids? ______
- 3. Is the following sentence true or false? Conduction is faster in metals than in other solids because metals have free electrons that transfer thermal energy.
- 4. Circle the letter of each sentence that is true about conduction.
 - a. Thermal energy is transferred without transfer of matter.
 - b. Matter is transferred great distances during conduction.
 - c. Conduction can occur between materials that are not touching.
 - d. In most solids, conduction takes place as particles vibrate in place.
- 5. Complete the table about conduction.

Conduction					
Type of Material	Quality of Conduction	Two Examples			
	Conducts thermal energy well	Copper;			
Thermal insulator		Wood;			

absolute zero. _____

18. Is the following sentence true or false? Matter can be cooled to